

Progress toward the Total Synthesis
of Bielschowskysin

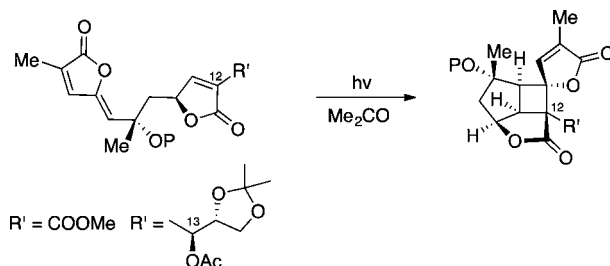
Steven D. Townsend and Gary A. Sulikowski*

Department of Chemistry, Institute of Chemical Biology, Vanderbilt University,
Nashville, Tennessee 37235-1822, United States

gary.a.sulikowski@vanderbilt.edu

Received August 20, 2013

ABSTRACT



Progress toward the total synthesis of bielschowskysin is described including introduction of the quaternary C12 and neighboring C13 stereocenters.

In 2004 Rodriguez and co-workers reported the structure of bielschowskysin, a diterpene isolated from the Caribbean gorgonian octocoral *Pseudopterogorgia kallos*.^{1,2} In addition to possessing a compelling structure for total synthesis studies, bielschowskysin was reported to exhibit antiparasitic activity and selective in vitro cytotoxicity against small-cell lung and renal cancer cell lines. This complex marine diterpene has proven to be a formidable target for total synthesis as its structural features include a highly oxygenated polycyclic ring system containing 11 stereocenters. Of particular interest is an almost fully substituted, central cyclobutane ring that incorporates a spirocyclic center. While a total synthesis of bielschowskysin has remained elusive, multiple groups have described various approaches toward its core ring system.³

(1) Marrero, J.; Rodriguez, A. D.; Baran, P.; Raptis, R. G.; Sanchez, J. A.; Ortega-Barria, E.; Capson, T. L. *Org. Lett.* **2004**, *6*, 1661–1664.

(2) Reviews on furanocembrane diterpenes: (a) Roethle, P. A.; Trauner, D. *Nat. Prod. Rep.* **2008**, *25*, 298–317. (b) Li, Y.; Pattenden, G. *Nat. Prod. Rep.* **2011**, *28*, 1269–1310.

(3) (a) Doroh, B.; Sulikowski, G. *Org. Lett.* **2006**, *8*, 903–906. (b) Miao, R.; Gramani, S. G.; Lear, M. J. *Tetrahedron Lett.* **2009**, *50*, 1731–1733. (c) Nicolaou, K. C.; Adsool, V. A.; Hale, C. R. H. *Angew. Chem., Int. Ed.* **2011**, *50*, 5149–5152. (d) Farcet, J.-B.; Himmelbauer, M.; Mulzer, J. *Org. Lett.* **2012**, *14*, 2195–2197. (e) Jana, A.; Mondal, S.; Hossain, M. F.; Ghosh, S. *Tetrahedron Lett.* **2012**, *53*, 6830–6833. (f) Himmelbauer, M.; Farcet, J.-B.; Gagnepain, J.; Mulzer, J. *Org. Lett.* **2013**, *15*, 3098–3101. (g) Saitman, A.; Sullivan, S. D. E.; Theodorakis, E. A. *Tetrahedron Lett.* **2013**, *54*, 1612–1615. (h) Meyer, M. E.; Phillips, J. H.; Ferreira, E. M.; Stoltz, B. M. *Tetrahedron* **2013**, *69*, 7627–7635.

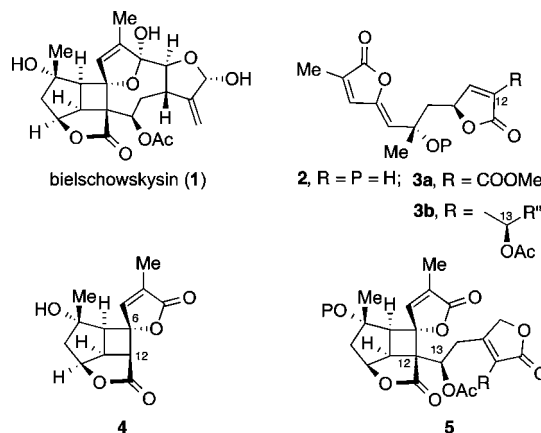
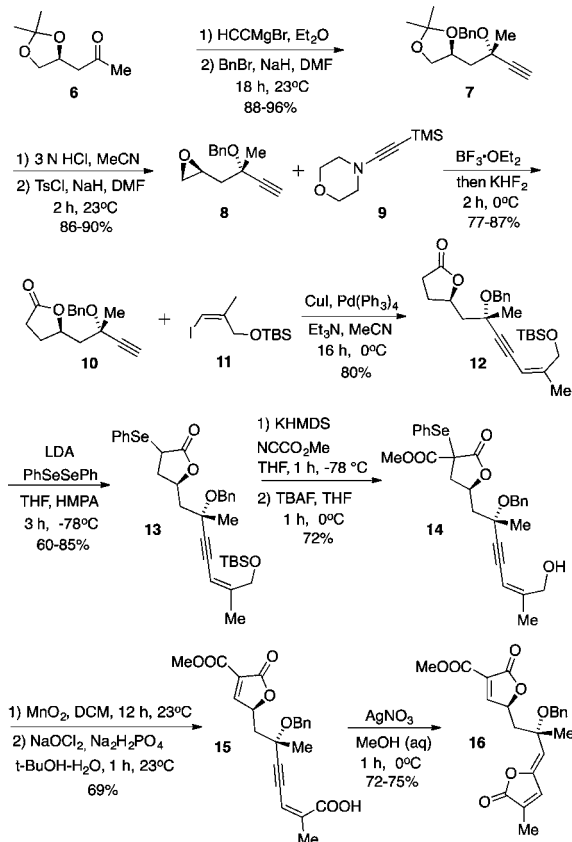


Figure 1. Structure of bielschowskysin, reported [2 + 2]-photocycloaddition,^{3a} and current synthetic analysis.

In 2006 our lab reported an intramolecular [2 + 2]-photocycloaddition upon irradiation of an acetone solution of bis-butenolide **2** to produce the Western half of bielschowskysin (**4**) including 5 of the 13 stereocenters.^{3a} Notably, this approach demonstrated stereocontrolled introduction of the tetrasubstituted C6 stereocenter of the desired configuration regardless of starting alkylidene

butenolide geometry. Since this communication, much of our effort has been directed toward introduction of the remaining elements of the bielschowskysin core structure in the form of butenolide **2**. Initial attempts to directly deprotonate C12 of photoadduct **4**, to react the resulting anion with an aldehyde, were unsuccessful. This is not surprising as the lactone is resistant to enolate formation, since deprotonation would result in the formation of a highly strained bridgehead olefin.⁴ To circumvent this problem, we turned our attention toward preparation and examination of the C12 substituted photochemical substrates **3a–b**. This strategy would involve formation of the C12–C13 bond *before* the photocycloaddition, avoiding the unmanageable task of C12 alkylation. Substrate **3b** would have the added advantage and challenge of incorporating the C13 stereocenter.

Scheme 1. Synthesis of Methyl Ester 16



We first examined the preparation of ester **3a** starting from methyl ketone **6**, available in four steps from (–)-malic acid.^{3a} As demonstrated earlier,^{3a} chelation-controlled addition of ethynylmagnesium bromide to ketone **6** proceeded with good stereoselectivity (ca. 5:1). Benzylolation of the resulting tertiary alcohol afforded ether **7** in 88–96% yield over two steps (Scheme 1). Acetonide removal released an intermediate diol that underwent a one-pot sequence of

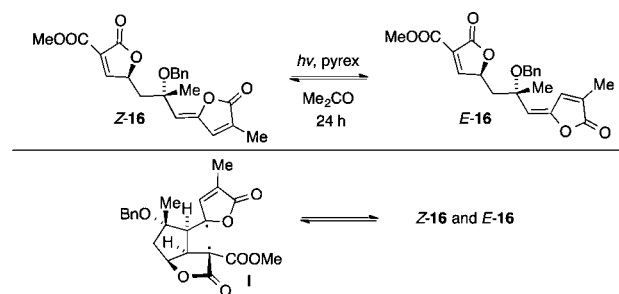
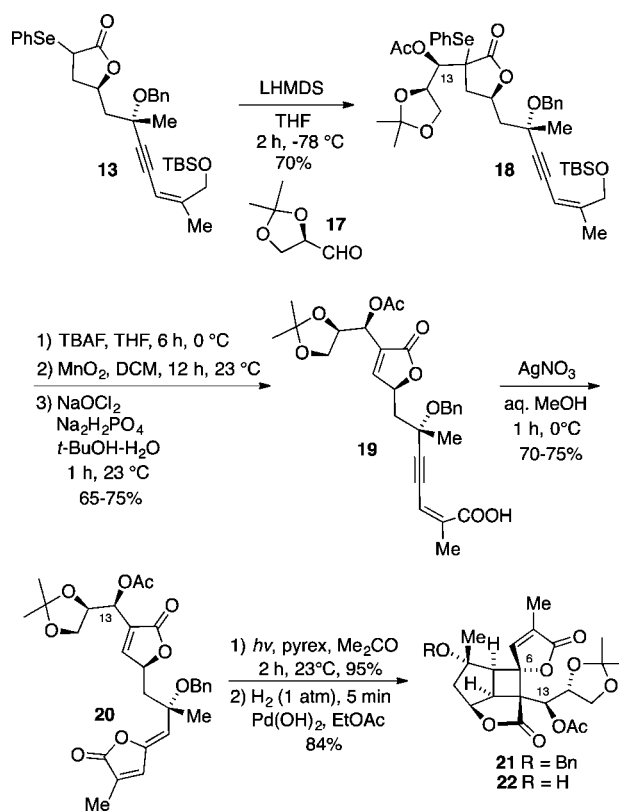


Figure 2. 1,4-Biradical intermediate **I** preferably reverts to ground state alkenes.

Scheme 2. Synthesis of Acetate 21



monotosylation and epoxide formation to give **8**.⁵ Next, the terminal epoxide was reacted with ynamine **9** leading to **10** following the Movassaghi and Jacobsen protocol.⁶ The lactone was extended through Sonagashira cross-coupling of its alkyne with vinyl iodide **11**, providing enyne **12** in 91% yield. Lactone **12** was further advanced to selenide **14** by a standard series of deprotonations, electrophilic quenches, and TBS removal. Sequential manganese dioxide and sodium hypochlorite oxidations served to convert the allylic alcohol to a carboxylic acid. Conveniently, the second

(4) Hayes, C. J.; Simpkins, N. S.; Kirk, D. T.; Mitchell, L.; Baudous, J.; Blake, A. J.; Wilson, C. *J. Am. Chem. Soc.* **2009**, *131*, 8196–8210.

(5) Cink, R. D.; Forsyth, C. J. *J. Org. Chem.* **1995**, *60*, 5746–5747.

(6) Movassaghi, M.; Jacobsen, E. *J. Am. Chem. Soc.* **2002**, *124*, 2456–2457.

oxidation was accompanied by selenoxide formation and elimination affording butenolide **15** in 69% yield. Finally, silver(I)-promoted cycloisomerization gave alkylidene butenolide **16**, completing the desired photochemical substrate.

Unlike our earlier photocycloaddition (**2** to **4**, Figure 1) irradiation of **Z-16** was unproductive, resulting only in double bond isomerization, producing a 1:1 mixture of **Z-16** and **E-16** (Figure 2). We hypothesized the carbomethoxy group stabilizes the intermediate 1,4-diradical **I** (Figure 2), rendering the intermediate diradical recalcitrant to formation of the strained cyclobutane, instead favoring reversion to alkylidene butenolides **Z-16** and **E-16**.⁷ To circumvent this effect, we reasoned it would be necessary to install the C13 oxygenation at a lower oxidation state. Our attention therefore turned to the assembly of the 13-acetoxy butenolide **20**, incorporating a key stereochemical element (C13) of bielschowskysin. To this end, substrate controlled addition of the enolate derived from lactone **13** to D-glyceraldehyde acetonide **17** resulted in adduct **18** following acetylation of the C13 hydroxyl group (Scheme 2).⁸ A series of reactions identical to those described in Scheme 1 afforded alkylidenebutenolide **20**. To our delight, irradiation of **20** provided photoadduct **21** in 95% isolated yield as a single isomer. In accord with the

previously published photocyclization of **2** to **4** (Figure 1) the C6 stereochemistry emerged with the oxygen positioned with the desired exo orientation to provide adduct **21**, establishing 7 of 11 stereocenters common to bielschowskysin. Finally, brief hydrogenation of **21** under 1 atm of hydrogen over Pearlman's catalyst resulted in removal of the benzyl protecting group to give alcohol **22**.

In closing, we have demonstrated stereocontrolled introduction of the C6 and C12 vicinal quaternary centers of bielschowskysin (**1**) via photocycloaddition of an appropriate C12-substituted bis-butenolide. While a carbonyl substituent prevents formation of the desired cyclobutane, adjustment of the C13 oxidation level to an alcohol allows for smooth conversion to the photoadduct. Utilizing its stereochemical preference, D-glyceraldehyde acetonide was used to set the stereochemistry at C13. Future efforts are directed toward advancing to butenolide **5** and ultimately bielschowskysin.

Acknowledgment. This work was supported by a grant from the National Science Foundation (CHE-1148569). S.T. gratefully acknowledges support from UNCF-Merck and Pfizer.

Supporting Information Available. Experimental procedures and full spectroscopic data for all new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

The authors declare no competing financial interest.

(7) Weixler, R.; Hehn, J. P.; Bach, T. *J. Org. Chem.* **2011**, *76*, 5924–5935.

(8) (a) Heathcock, C. H.; Young, S. D.; Hagen, J. P.; Pirrung, M. C.; White, C. T.; VanDerveer, D. *J. Org. Chem.* **1980**, *45*, 3846–3856. (b) Heathcock, C. H.; White, C. T.; Morrison, J. J.; VanDerveer, D. *J. Org. Chem.* **1981**, *46*, 1296–1309.